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Subject

Basic electro mechanical engineering

Submitted

to Sir Ashraf Ali

# Question #1 Part A

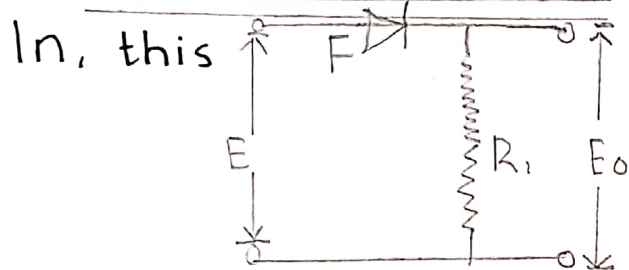
P1

Answer:

Diode:-

A diode is an electronic component that has two terminals, limits current to one direction. diodes have anode and Cathode. Positive current Normally flows from anode to Cathode. diodes are useful for Protecting circuitry from harmful Voltage or current. diodes are a basic building block of charge-collecting element in many detector.

## Half-Wave Rectifier



→ one diode or one Semi-conductor diode is used.

→ ordinary transformer is used

→ It converts half cycle of applied A.C. signal into D.C.

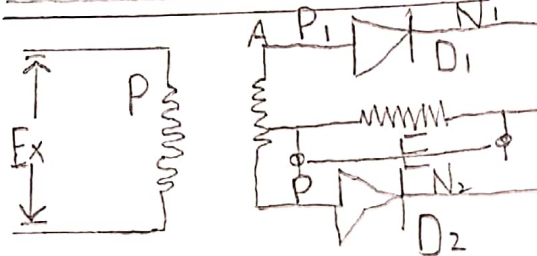
→ The Value of  $I_{ms} = \frac{I_o}{2}$ .

→  $I_{dc} = \frac{I_o}{\pi}$

→ The Value of ripple factor

$$r = \sqrt{\left(\frac{I_{ms}}{I_{dc}}\right)^2} - 1 = 121\%$$

## Full Wave Rectifier



In this two diodes or one double diode or two Junction are used.

Centre tap transformer is used.

It converts the whole cycle of applied A.C. signal into D.C.

$$I_{ms} = \frac{I_o}{\sqrt{2}}$$

$$I_{dc} = \frac{2I_o}{\pi}$$

The Value of  $r$  in it is 48.2%.

Question # 1 Part B

P 2

Solution:

Required: Find temperature of mixture

$$- \text{Lose}_{\text{heat}} = \text{Gain}_{\text{heat}}$$

$$- [C_{\text{Au}} (\text{mass}) (\Delta T)] = (C_{\text{H}_2\text{O}}) (\text{mass}) (\Delta T)$$

$$- [(0.129 \text{ J/g} \cdot ^\circ\text{C}) (97 \text{ g}) (T_F - 785^\circ\text{C})] = [(4.184 \text{ J/g} \cdot ^\circ\text{C}) (323) (T_F - 15^\circ\text{C})]$$

$$- [(12.5) (T_F - 785^\circ\text{C})] = (1.35 \times 10^3) (T_F - 15^\circ\text{C})$$

$$- 12.5 T_F + 9.82 \times 10^3 = 1.35 \times 10^3 T_F - 2.02 \times 10^4$$

$$3 \times 10^4 = 1.36 \times 10^3 T_F$$

$$T_F = 22.1^\circ\text{C}$$

Answer.

Answer:

Isobaric Process:•

An Isobaric process is a thermodynamics process in which the Pressure stays constant  $\Delta P = 0$ . the heat transferred to the System does work, but also change the internal energy of the System. using this convention, by the first law of thermodynamics, where  $W$  is work,  $U$  is internal energy, and  $Q$  is heat.

IsoMetric Process:•

An Isochoric process, also called a constant-volume process an isovolumetric process or an Isometric process is a thermodynamic process during which the Volume of the closed System undergoing such a process remains constant.

Isothermal Process:•

An Isothermal Process is a thermodynamic process in which the temperature of a System remains constant. the transfer of heat into or out of the System happens so slowly that thermal equilibrium is maintained.

## Adiabatic Process:-

P4

An adiabatic process occurs without transferring heat or mass between a thermodynamic system and its surrounding. unlike an isothermal process, an adiabatic process transfer energy to the surroundings only as work.

## Question #2 Part B

P5

Answer:

Solution: -

$$e = 1 - \frac{T_c}{T_H}$$

$$e = 1 - \frac{300\text{K}}{500\text{K}}$$

$$e = 40\%$$

$$\text{Actual } e = 0.5e_i = 20\%$$

$$e = \frac{W}{Q_H}$$

$$W = eQ_H = 0.20(600\text{J})$$

$$\text{Work} = 120\text{J}$$



## Question # 3 Part A

P6

Answer:

### Internal Combustion engine

→ Name it self Says that Combustion take Place inside the cylinder.

→ Temperature is higher.

→ Pressure is higher.

→ In Ic engine, piston and Connecting rod is use.

→ efficiency is higher.

→ lighter in weight.

→ Ic engine is costly.  
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→ less time required to start.

→ Pressure generated inside the engine is due to combustion of fuel.

→ Fuel tank required to Store fuel.

### External Combustion engine

Name it self Says that, Combustion take place outside the cylinder.

Temperature is lower.

Pressure is lower.

In Ec engine, Stuffing box is use.

efficiency is lower.

Heavy in weight.

Ec engine is cheaper compared to Ic engine.

More time required to start.

Pressure generated inside the engine is due to Steam of water.

Boiler and water storage required to generate Steam.

# Question # 3 Part B

P7

Answer:-

Solution:-

$$\Delta Q = 0$$

$\Rightarrow$  Find  $P_B$

$$P_A V_A = P_B V_B$$

$$P_B = P_A \left( \frac{V_A}{V_B} \right)^{\gamma}$$

$$P_B = 32.4 \text{ ecfm}$$

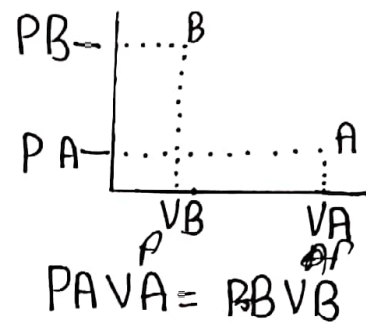
$$\boxed{\text{or } 3284 \text{ KPA}}$$

$\Rightarrow$  Find  $T_B$

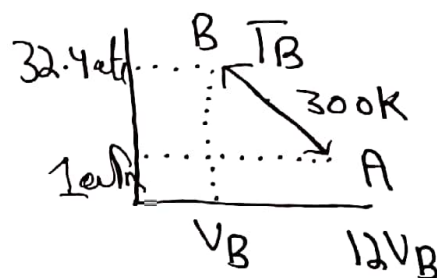
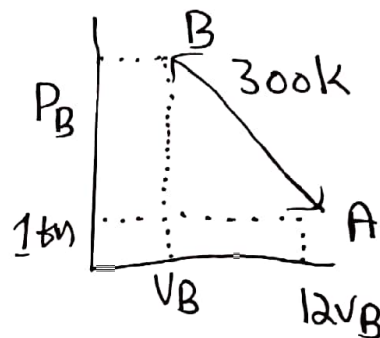
$$\frac{P_A V_A}{T_A} = \frac{P_B V_B}{T_B}$$

$$\Rightarrow \frac{(1 \times 12)}{300 \text{ K}} = \frac{(32.4)(1)}{T_B}$$

$$\boxed{T_B = 810 \text{ K}}$$



$$\frac{P_A V_A}{T_A} = \frac{P_B V_B}{T_B}$$





Answer:

### Conduction

→ In conduction, the heat transfer takes place between objects by direct contact.

→ The heat transfer takes place due to the difference in temperature.

→ The heat transfer in conduction is slow.

→ The heat transfer occurs through a heated solid object.

→ It does not follow the law of reflection and refraction.

### Convection

In convection, the heat transfer takes within the fluid.

The heat transfer due to the difference in density.

The heat transfer in convection is faster.

The heat transfer occurs through intermediate objects. For example heat transfer between air and water.

It does not follow the law of reflection and refraction.

# Question # 4 Part B

P9

Answer:

Solution:-

$$\Rightarrow Q_{\text{water}} = Q_{\text{pb}}$$

$$\Rightarrow m_{\text{water}} c_{\text{water}} \Delta T_{\text{water}} = - (m_{\text{pb}} c_{\text{pb}} \Delta T_{\text{pb}})$$

$$\Rightarrow 125 (4.18) (T_f - 23) = -75 (0.13) (T_f - 435)$$

$$\Rightarrow 522.5 T_f - 12017.5 = -9.75 T_f + 4241.25$$

$$\Rightarrow + 9.75 T_f + 12017.5 = +9.75 T_f + 4241.25$$

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$$582.25 T_f = 16258.75$$

$$T_f = 30.5^\circ\text{C}$$

Answer: .

Solution: .

→ Nothing that heat transfer through the roof is by conduction and the area of the roof is

$$A = 6\text{m} \times 8\text{m} = 48\text{m}^2$$

The steady rate of heat transfer through the roof is determined to be,

$$Q = KA(T_1 - T_2)/L = (0.8)(48)(25 - 0)/0.25$$
$$= 384\text{W} = 3.84\text{kW}$$

→ The amount of heat cost through the roof during at 10-hours period and its cost are determined from

$$Q = Q\Delta t = (3.84\text{kW})(10\text{h}) = 38.4\text{kWh}$$

$$\text{Cost/day} = (\text{Amount of energy})(\text{unit cost of energy}).$$

$$\Rightarrow (3.84\text{kWh})(\cancel{2} \ 0.2/1\text{kWh}) = 87.68$$

$$\text{Cost/month} = \text{cost/day} \times 30\text{ days}$$
$$= 7.68 \times 30 = 8230.4$$