

ID: 13748

Name Hina Ishaq

①

Thermodynamics

Q 1(a)

i) The Ideal Gas Law:

The gas constant (R) is a mathematical combination of all the individual gas law constants (c_b, c_c, c_g, c_a)

$$\frac{PV}{nT} = R$$

The ideal gas law is more commonly written as

$$PV = nRT$$

The value of R is $\left[\frac{0.0821 \text{ L}\cdot\text{atm}}{\text{K}\cdot\text{mol}} \right]$

$$PV = nRT$$

When using this equation you must have the following units:

- Pressure = atm
- Volume = L
- Temperature = K

ID: 13748

Name: Hamza Ishaq (2)

Thermodynamics

ii) Dalton's Law of Partial Pressure: states that the total pressure of a mixture of gases is the sum of the partial pressures of its components.

The partial pressure of a gas in a mixture is the pressure that the gas would exert if alone.

Mathematically:

$$P_T = P_A + P_B + P_C$$

II) 13748

Name Hamza Ishaq

(3)

Thermodynamics

Q1 (b)

Solution:

$$P_1 V_1 = P_2 V_2$$

$$\frac{P_1 V_1}{V_2} = \frac{P_2 V_2}{V_1}$$

$$P_1 = 4 \text{ atm}$$

$$P_2 = ?$$

$$V_1 = 6 \text{ litres}$$

$$V_2 = 2.50$$

$$P_2 = \frac{P_1 V_1}{V_2}$$

$$= \frac{(4 \text{ atm})(6 \text{ L})}{(2.50 \text{ L})}$$

$$\boxed{P_2 = 9.6 \text{ atm}}$$

ID: 13748

(4)

Name: Hange Ishaq

Thermodynamics.

Q 2 (a)

Answer:

Key factors of ideal refrigerator:

An ideal refrigerant should have the following properties.

1. Low viscosity
2. Low freezing point
3. Low boiling point
4. Low heat capacity
5. Low specific volume
6. Low saturation pressure
7. Odorless
8. High latent heat of vaporization
9. Good thermal conductivity.
10. High COP
11. Non inflammable and non explosive.
12. High critical pressure and temp.

ID 13748

Name: Hamza Ishaq

Thermodynamics

(5)

Q 2(b)

Vapour Absorption Refrigeration:

Principle

Here the heat energy is utilized to achieve the refrigeration. An electric heater or steam is used to add the heat to the refrigerant for its evaporation. Also an absorber, a pump and a generator are used to complete the cycle.

Construction

Here a throttle valve is connected between an evaporator and a condenser. One pump is connected between an absorber and a separator. Water is filled in the absorber and it is connected to the evaporator. The evaporator is kept in the storage room. The separator is connected to the condenser. Both the condenser and evaporator consists of pipes in coil form to provide more

ID 13748

Name: Hamza Ishaq

(6)

Thermodynamics

more contact surface area for the refrigerant. Arguments are made to circulate the cold water around the condenser and in the absorber. An electric heater is housed in the separator. Tray may be positioned to collect the exit water near by the absorber and condenser. A receiver is connected in b/w condenser and the throttle valve.

Working:

Dry ammonia vapour from the evaporator enters the absorber containing water where it is absorbed by the water becomes a strong ammonia solution with an increase in temperature. The heat generated during this process is removed to some extent by circulating cold water through a pipe. It is then driven out from the solution to the condenser where it is condensed and return to the liquid state. The liquid ammonia is then collected in the receiver. The

ID: 13748

Name: Hamza Ishaq

(7)

Thermodynamics.

high pressure liquid ammonia is then passed through the throttle valve when it is expanded with decrease in temperature and pressure. Later it enters the evaporator kept in the storage room.

Comparison b/w Vapour Compression and Vapour absorption refrigeration.

Vapour Compression

Vapour absorption

- | | |
|---|--|
| → Works using mechanical energy | → Works using heat energy |
| → Refrigeration capacity is less than 1000 tons | → Refrigeration capacity is greater than 1000 tons |
| → COP is much higher | → COP is less |
| → Noisy due to compressor | → Pump noise is less |
| → chances of leakage of refrigerant are more | → No leakage |
| → maintenance and operating cost are high | → less |
| → Smaller in size | → larger |
| → Wear and tear are more | → wear and tear are less. |

ID: 13748

Name: Hamza Ishaq

⑧

Thermodynamics

Q3

Difference b/w Fire tube and Water tube Boilers

Fire tube Boiler

- Hot flue gases flow inside the tube and the water
- These boilers are generally internally fired
- The boiler pressure limited to 20 bar
- The fire-tube boiler has a lower rate of steam production
- This boiler is difficult to construct
- It occupies large floor space for a given power

Water tube Boiler

- water flows inside the tube and hot flue gases outside the tube
- These boilers are generally extra really fired
- The boiler pressure is limited up to 100 bar.
- A higher rate of steam production.
- Simple in construction.
- It occupies small floor space

ID: 13748

Name Hamza Ishaq

(9)

Thermodynamics

- They are difficult to separate and clean as they are internally fixed → They are easy to separate and clean as they are externally fixed
- They require a large shell diameter. Because the fire-tube is situated inside the shell → They require a small shell diameter.

ID: 13748

Name: Hamza Ishaq

Thermodynamics

(10)

Q4

Answer.

Stroke: A stroke is movement of the piston from top dead centre (T.D.C) to bottom dead centre (B.D.C) or from (B.D.C) to (T.D.C)

Cycle: It is a series of event that repeat themselves.

Four Stroke Engine:

Four stroke cycle engine works on four stroke principle i.e there are four strokes in one cycle of such engine. Four-stroke engine is also called four cycle engine or "Otto-cycle engine".

IID: 13748

Name: Hamza Ishaq

Thermodynamics

(11)

The four strokes of otto-cycle engine are

- ① Intake stroke
- ② Compression stroke
- ③ Power stroke
- ④ Exhaust stroke

Intake stroke:

On the intake stroke, the piston is moving down due to which a partial vacuum is produced inside the cylinder. The intake valve is open, therefore, atmospheric pressure pushes the fresh air fuel mixture in the cylinders through the intake valve.

Compression stroke:

When the piston reaches the B.D.C on the intake stroke, the intake valve closes then the piston moves up on the compression stroke. During this stroke both

ID: 13748

Name: Hamza Ishaq

Thermodynamics

(12)

The valves are closed, therefore no air-fuel mixture can enter or goes out of the cylinder. Therefore when the piston moves up the mixture inside the cylinder is compressed.

In modern engine it is compressed to one-eighth or one-ninth of the original volume. The amount to which the air-fuel mixture is compressed is called "the compression ratio".

If the mixture is compressed to one-eighth of the original volume, the compression ratio is '8' to '1' (8:1)

Power stroke:

During the compression stroke, the piston moves up when it gets near T.D.C a spark which ignites the compressed air-fuel mixture. Due to which the temperature and the pressure of gas rises. The pressure

ID: 13748

Name: Homze Ishaq

Thermodynamics

(13)

becomes about 600 Psi, which push the piston down. As during this stroke, we obtain power therefore it is called power stroke.

The piston transmit this power through the connecting rod to the crank due to which the crankshaft rotates this rotary motion is carried through shaft and gears to the car wheels due to which wheel turns.

Exhaust Stroke.

During the power stroke at the piston reaches B.D.C, the exhaust valve opens. Then the piston moves up on exhaust stroke when the piston moves up it pushes out the burnt gases from the cylinder through exhaust valve. When the piston reaches T.D.C the exhaust valve closes and intake valve opens for the second cycle.

ID: 13748

Name: Hariz Ishaq

(14)

Thermodynamics

Compression Ratio:

The volume in which piston reciprocates i.e the volume from B.D.C to T.D.C is called swept volume (V_s)

The volume in which fresh air-fuel mixture is compressed by the piston in the cylinder is called "clearance volume"

Thus the total volume of the cylinder is the sum of the swept volume and clearance volume

$$\text{i.e } V_T = V_s + V_c$$

And the compression ratio is the ratio of the total volume to the clearance volume.

$$\text{i.e compression ratio} = \frac{\text{Total Volume}}{\text{Clearance Volume}} = \frac{V_T}{V_c}$$

ID: 13748

Name: Hamza Ishaq

Thermodynamics

(15)

Q5 (a)

Concept of Scavenging:

In an internal combustion engine scavenging is the process of replacing the exhaust gas in a cylinder with the fresh air/fuel mixture (or fresh air, in the case of direct-injection engines) for the next cycle. If scavenging is "complete", the remaining exhaust gases can cause improper combustion for the next cycle, leading to reduced power output.

Scavenging is equally important for both two-stroke engines and four-stroke engines. Most modern four-stroke engines use crossflow cylinder heads and valve timing overlap to scavenge the cylinders. Modern two-stroke engines use either Schnoroide scavenging (also known as "loop scavenging") or uniflow scavenging.

ID: 13748

Name: Hamza Ishaq.

Thermodynamics (16)

Q 5 (b)

Answer

The Rickshaw due to being equipped with 2-stroke engines, are the most inefficient vehicles in complete burning of fuel and thus contribute most to emission of air pollutants in the environment. The major pollutants from two-stroke engines are Carbon Monoxide (CO), Nitrogen oxides (NO_x), Hydrocarbons (HC) and particulate matter (PM). Their presence in the environment causes a number of respiratory diseases and other illnesses. For example, CO and NO_x are notorious irritants of respiratory system and have potent suffocating action. PM causes premature death, and illness. Its presence is accompanied by increased hospital admissions for asthma and other bronchial conditions such as bronchitis etc.

ID: 13748

Name: Hamza Ishaq (17)

Thermodynamics

None of the locally assembled rickshaws meet standards fixed by Pakistan standard quality control Authority (PSQCA) for smoke emission. According to the PSQCA code, two-stroke exhaust contain 45 Per cent and old two-stroke has six Per cent carbon mono oxide.

