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Q No 1

P. No (a)

Ans:Diode:-

A diode is an electronic component that:  
has two terminals;  
limits current to one  
dimension direction.

→ Diodes have an anode and a cathode.

→ Positive current normally flows from the anode to the cathode.

→ Diodes are a basic building block of the charge-collecting

element in many detectors.

Half wave

→ Half wave rectifier is a low-efficiency rectifier.

→ Half wave rectifier does not require center tapping of the secondary winding of transformer.

→ Half wave rectifier less electronic components as full wave rectifier.

→ Half wave less costly than full wave is easily.

→ one diode conducts in each half cycle of input.

Full wave rectifier is a high efficiency rectifier.

Full wave ~~rectifier~~ <sup>rectifier</sup> require center tapping of the secondary winding of the transformer.

Full wave ~~rectifier~~ <sup>rectifier</sup> more electronic components.

Full wave rectifier

Two diode conduct in each half cycle of input.

Prob (b)

Solution:

$$-\text{lose heat} = \text{Gain heat}$$

$$- [C_{Cu} (\text{mass}) (\Delta T)] = (C_{H_2O} (\text{mass}) (\Delta T))$$

$$- [(0.129 \text{ J/g}^\circ\text{C}) (979) (T_0 - 785^\circ\text{C})] = (4.180 \text{ J/g}^\circ\text{C}) (3239) (T_0 - 15^\circ\text{C})$$

$$- [12.5 (T_0 - 785^\circ\text{C})] = (1.35 \times 10^3) (T_0 - 15^\circ\text{C})$$

$$- 12.5 T_0 + 9.82 \times 10^3 = 1.35 \times 10^3 T_0 - 2.02 \times 10^4$$

$$3 \times 10^4 = 1.36 \times 10^3 T_0$$

$$T_0 = 22.1^\circ\text{C} \quad \text{Ans}$$

Ques

5

Solution:

Nothing that heat transfer through the rods is by conduction and the area of the rods is:

$$A = 6m \times 8m = 48m$$

The steady rate of heat transfer through the rods is determined to be

$$Q = kA(T_1 - T_2)/L = (0.8)(48)/(0.5-0)/0.25$$

$$= 3840 \text{ W}$$

$$= 3.84 \text{ kW}$$

The amount of heat lost through the rods during a 20 hour period and its cost are determined from:

$$Q = \dot{Q} \cdot \Delta t = (3.84 \text{ kW})(20 \text{ h})$$

$$Q = 38.4 \text{ kWh}$$

$$\text{cost/day} = (\text{Amount of energy}) \cdot (\text{unit cost of energy})$$

$$= (38.4 \text{ kWh})(\$0.2/\text{kWh})$$

$$= \$7.68$$

$$\text{cost/day} = \$7.68$$

$$\text{cost/month} = (\text{cost/day}) \times (30 \text{ day/month})$$

$$= \$7.68 \times 30$$

$$\text{cost/month} = \$230.4$$

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Q No 4

Q No (a)

Ans:

	Conduction	Convection
→	Energy transferred by direct contact.	occurs in gases and liquids.
→	Energy flows directly from warmer to cooler objects.	Movement of large number of particles in same direction.
→	Continues until object temperature are equal	cycle occurs while temperature differences exist.

P.No(6)

Solution:

$$Q_{water} = - Q_{pb}$$

$$m_{water} C_{water} \Delta T_{water} = - (m_{pb} C_{pb} \Delta T_{pb})$$

$$125 (4.18) (T_b - 25) = -75 (0.13) (T_b - 435)$$

$$522.5 T_b - 12017.5 = -9.75 T_b + 42412.5$$

$$+9.75 T_b + 12017.5 = +9.75 T_b + 42412.5$$

$$532.25 T_b = 16258.75$$

$$T_b = 30.5^{\circ}C \quad \underline{\text{Ans}}$$

P.No(7):

P.No(8)

Solution:

$$e = 1 - \frac{T_c}{T_H}$$

$T_H$

$$e = 1 - \frac{300K}{800K}$$

$800K$

$$e = 40\%$$

page (7)

$$\text{Actual } e = 0.20 = 20\%$$

$$e = \frac{W}{Q_H}$$

$$W = e Q_H$$
$$= 0.20 (600 \text{ J})$$

$$\boxed{\text{work} = 120 \text{ J}}$$

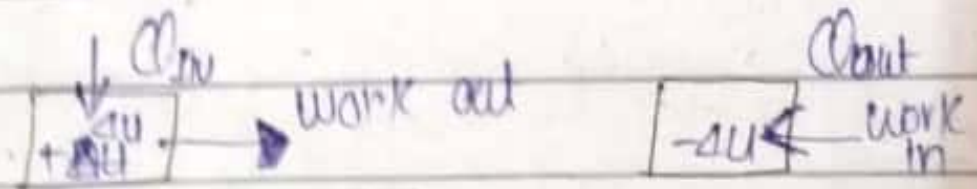
P.1000)

Ans

### ISOBARIC PROCESS

constant pressure  $\Delta P = 0$

$$\Delta Q = \Delta U + \Delta W \quad \text{But } \Delta W = P \Delta V$$



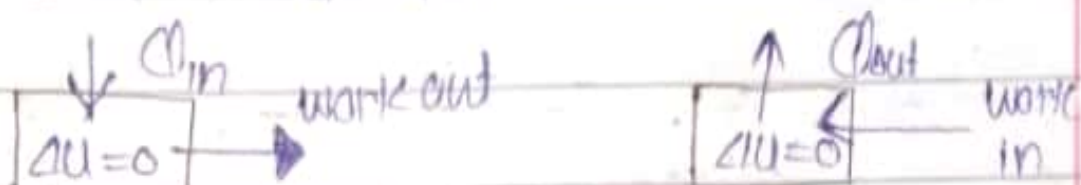
Heat In = Work out + Increase in internal energy.

Heat Out = Work out + Decrease in internal Energy.

## ISOTHERMAL PROCESS

Const. Temperature,  $\Delta T = 0$ ,  $\Delta U = 0$

$$\Delta Q = \Delta U + \Delta W \quad \text{And} \quad \Delta Q = \Delta W$$



Net heat input = - Work output

Work Input = Net heat out

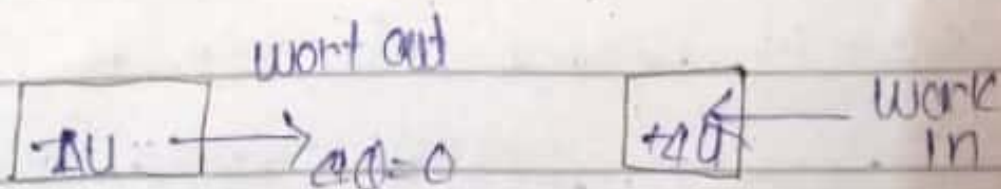
## ADIABATIC PROCESS

No Heat Exchange,  $\Delta Q = 0$

$$\Delta Q = \Delta U + \Delta W; \quad \Delta W = -\Delta U \quad \text{or} \quad \Delta U = -\Delta W$$

$$\Delta W = -\Delta U$$

$$\Delta U = -\Delta W$$

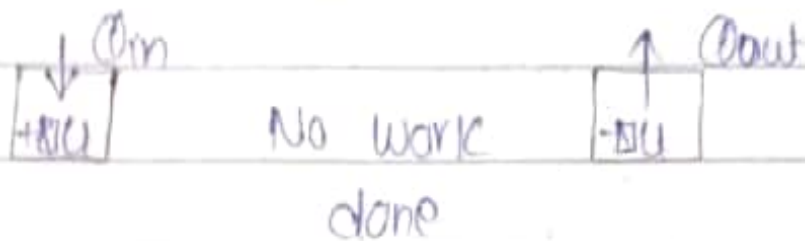


work done at EXPENSE of  
internal energy INPUT work  
INCREASES internal energy.



# ISOCHORIC PROCESS

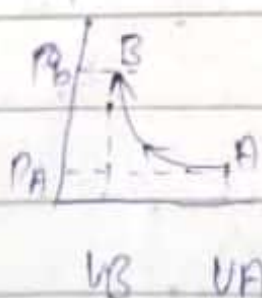
constant volume,  $\Delta V = 0$   $\Delta W = 0$   
 $\Delta Q = \Delta U + \Delta W^0$  so that  $\Delta Q = \Delta U$



Heat in = Increase in internal energy  
 Heat out = Decrease in internal energy.

Q No 3  
 P. No (b)

Solution:

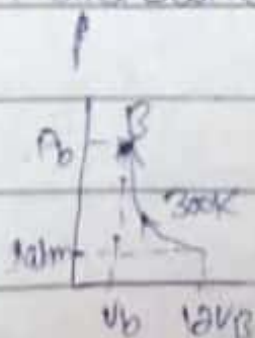


$$P_A V_A = P_B V_B$$

$$\textcircled{a} P_A V_A = P_B V_B$$

$$T_A \quad T_B$$

Adiabatic (cont): Find  $P_B$



$$P_A V_A = P_B V_B$$

solve for  $P_B$ :

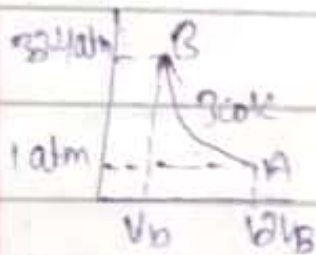
$$P_B = P_A \left( \frac{V_A}{V_B} \right)^\gamma$$

$$P_b = P_a \left( \frac{V_a}{V_b} \right)^{\gamma}$$

$$P_b = (1 \text{ atm}) (12)^{1.4}$$

$$P_b = 32.4 \text{ atm or } 3284 \text{ kPa}$$

Adiabatic (const.): Find  $T_b$



$$P_A V_A^\gamma = P_B V_B^\gamma$$

Solve for  $T_b$

$$(1 \text{ atm}) (12 \text{ V})^\gamma = (32.4 \text{ atm}) (V_b)^\gamma$$

$$(300 \text{ K})$$

$$T_b = 810 \text{ K}$$

Q. No 3

P. No (9)

Ans: Internal combustion engine:

The commercially successful combustion engine was created by F. Lenoir in 1860. And then more modified in 1876 by N. Otto.

External combustion engine:-

The external combustion engine is working as heat engine where a working as liquid or gas.

diff b/w External and Internal engine:

The difference b/w external and internal combustion engine is quite straight forward and made obvious by the difference

In their names, in External combustion engine the fuel is not burned inside the engine.

While in internal combustion engine the combustion chamber lies right in the middle of the engine.

ECF working based that heated while ICE within the engine to produce work.